BATTERY MATERIAL QUALITY: THE IMPORTANCE OF PHYSICAL PROPERTIES

Anthony Chalou

Head of Market Development, Anton Paar

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COVALENT ACADEMY

Industrial Applications of Advanced Metrology Episode 37



Modern, digitally-empowered analytical services platform delivering quality data and expert analysis to accelerate advanced materials and device innovation.



Comprehensive Solutions Stack

50+ Cutting-edge instruments in-house, 150+ Techniques

Analytical Services

Advanced Modeling

Method Development

Custom Consulting Solutions



High-touch, High-Quality Services

Ionic Membership Program

Enterprise Metrology Solutions

Instant, Secure Access to Data and Reports

Expanding Toolkit in Custom Digital Platform



Flexible Business Models

LiveViewTM (real-time collaboration)

Co-op and Tool-share Opportunities

Training and Certification on Instrumentation

Laboratory Audits



Rich Network of Partnerships

Partner to World's Leading Instrument Manufacturers and Labs

Expanding access to Advanced Instruments and Analysis Tools

Lab Connections and Applications Learning



Who We Are, Who We Serve

50+ People, 14 PhDs

Comprehensive, Modern Analytical Capabilities

\$20M +State-of-art Lab in Sunnyvale, CA

600+ Clients, 15-30 new clients / week

Covalent's Analytical Services & Technical Groups





EBIC / OBIC failure analysis

Electron Microscopy



- FIB-SEM & HR-SEM with EDS; EBSD; 3D
- Lamella Preparation incl. specialized lift-outs



Failure Analysis

- Root-Cause Failure Analysis
- DPA / Mechanical X-section
- Dve & Prv Test
- Hot Spot Detection
- Emission Microscopy
- NIR / IR Imaging

Surfaces Microscopy and

Chemicals,

Materials,



Microscopy & Profilometry

- Chromatic Aberration
- Digital Optical Microscopy
- Laser Scanning Confocal
- White Light Interferometry
- Scanning Acoustic Microscopy (SAM)



Mechanical Testing

- AFM & Advanced AFM Modes (EFM, KPFM, MFM, PFM, PiFM)
- Nano-indent / Nano-scratch
- Rheometry / Viscosity
- DMA / TMA (bend/stretch/compression)
- Tensile testina



Analytical Chemistry

- Mass Spectroscopy: ICP-MS and LA-ICP-MS: GCMS
- ICP-OES / GDOES
- Raman
- NMR (solid / liquid + 1,2,3 nuclei)

Misc. Material Properties

Gas Adsorption / Chemisorption

Foam Density / Skeletal Density

Particle Analysis: DLS / ELS /

size distribution / zeta potential

Thermal Analysis: DSC, TGA

Surface Zeta Potential

/ Tap Density

Porometry / Pycnometry

- XPS, UPS, ISS
- SIMS, TOF-SIMS



X-ray Characterization

- X-Ray Diffraction (XRD)
- X-Ray Reflectometry (XRR)
- Micro-computed X-ray Tomography (Micro-CT)
- 2D / 2.5D / 3D X-ray Inspection & X-ray Radiography
- ED-XRF / WD-XRF



Optical Characterization

- Fourier Transformed Infrared Spectroscopy (FTIR and ATR-FTIR)
- Spectral Ellipsometry & Advanced Optical Modeling
- UV-Vis-NIR Spectroscopy

Covalent Partners





Partnership with Anton Paar announced in May, 2020

- Established the Anton Paar Demonstration Facility in Covalent's Silicon Valley Laboratory with goals of
 - Expanding industry access
 - Developing new analytical applications
- Later expanded partnership to deliver industry-leading porous materials and powders analysis
- Partners continue to collaborate in advanced applications development

- Anton Paar Instruments at Covalent Metrology include:
- SurPASS 3
- Litesizer 500
- MCR 702 Rheometer / DMA
- STeP 6 Nanoindentation platform
- Ultrapyc 5000 Micro
- Autosorb iQ C-XR-XR with CryoSync accessory
- Porometer 3G and DualAutoTap
- NEW Upgraded Nova 800 BET (Gas Adsorption) Analyzer

Other Covalent Partners



Introducing



Anthony Chalou

Head of Market Development, Anton Paar



- Anthony Chalou is the Head of Market Development at Anton Paar
- He holds two Masters degrees in Applied Analytical Chemistry and has more than 20 years of experience with analytical instrumentation, both in scientific applications as well as in sales and marketing
- His current focus is on the characterization of physical properties of materials used in the R&D and QC of lithiumion batteries





ANTON PAAR SOLUTIONS FOR BATTERY MATERIAL CHARACTERIZATION

Anthony Chalou Head of Market Development Anton Paar GmbH



OUTLINE

- > Main components
- > LIB Types
- > LIB Chemistries
- > Cell formats
- > Pros & cons of common formats
- > Battery trilemma
- > Emerging trends
- > Physical parameters of raw materials
- > Physical parameters of slurries
- > Questions



MAIN COMPONENTS OF A LITHIUM ION BATTERY



TYPES OF LITHIUM ION BATTERIES





maturity

- Excellent cycle life, and improved safety
- Suffers from low energy density

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LITHIUM-BASED RECHARGEABLE SYSTEMS

Technology	Anode	Cathode
Lithium Cobalt Oxide 3.6V	Graphite –Gr/Si	LCO
Lithium Manganese Oxide Spinel 3.8V	Graphite –Gr/Si	LMO
Lithium Nickel Manganese Cobalt 3.7-3.8V	Graphite –Gr/Si	LNMC (111,532, 622, 811)
Lithium Nickel Cobalt Aluminum 3.65V	Graphite –Gr/Si	LNCA
Lithium Iron Phosphate 3.2V	Graphite –Gr/Si	LFP*
Lithium Manganese Iron Phosphate 3.7V	Graphite –Gr/Si	LMFP*
Lithium Titanite Oxide 2.3/2.4V	LTO	NMC, LMO, LFP
Lithium Nickel-Rich Oxide 3.8V	Graphite –Gr/Si	eLNO*
Lithium Metal 3.7V	Lithium Metal	NMC
Lithium Nickel Manganese Oxide Spinel 3.8V (Mn Rich)**	Graphite –Gr/Si	LNMO*
Lithium Sulfur 2.4V**	Lithium Metal	Sulfur*

* Cobalt free ** R&D stage

MOST COMMON FORMATS OF BATTERY CELLS





A spirally wound design (jelly-roll). Designated by size, e.g. 18650 cylindrical battery (Diameter: 18.6 mm, length: 65.2 mm; code for cylindrical shape: 0) Prismatic

A prismatic design indicate a flat battery design. The stacks can be wound (as shown in the photo) or stacked (with alternating cathode/separator/anode structure). The stacks are usually inserted into rigid casing to form prismatic containers



Rather than rigid metallic casing, conductive foil-tabs are welded to the electrodes and seal the battery fully. The stacks inside can be wound or stacked. Swelling and gassing could be a concern for pouch cells.

Coin cell



Also known as button cell. The cells are stacked into a tube. Most coin cells are single-use.

Source: J.M. Tarascon, Nature 2001, Sanyo and Panasonic

PROS AND CONS OF THE DIFFERENT FORMATS OF BATTERY CELLS



Cylindrical		Prismatic		Pouch	
+	-	+	-	+	-
High energy density	Space inefficiency	High energy density	Mainly large and high-capacity cells	High volumetric energy density	Mechanical limitations
Wide operating voltage range	Low volumetric energy density (at pack level)	Wide operating voltage range	Requires special orders	Wide operating voltage range	Requires special orders
Wide power density range	Thick	Wide power density range	Thicker than pouch cells	Wide power density range	Swelling
Good cycle life (500-1000 cycles)	No custom sizes	Good cycle life (500-1000 cycles)	No standard sizing	Good cycle life (500-4000 cycles)	No internal safety devices
Low self-discharge (~1% / month)	Difficult thermal management	Low self-discharge (~1% / month)	Difficult thermal management	Low self-discharge (~1% / month)	Vulnerable to shock and vibration
Quick charge possible		Quick charge possible		Quick charge possible	
Internal safety devices		Internal safety devices		Better thermal management	
Standard sizes (no custom orders)		Customization possible for mass production		Easy to custom make	
Robust		Robust		Mass production for large cells	
Mass production		Thin (Better pack volume)		Thin (Better pack volume)	

THE "IDEAL" BATTERY





Fast-charging batteries have shorter life



EMERGING TRENDS



RECYCLING

- Black mass
- > Hydrometallurgy
- Pyrometallurgy
- > New regulations

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SOLID STATE BATTERIES

- Enhanced safety
- Extended lifespan
- Increased energy density
- > More expensive
- Challenges in scaling up



DRY COATING

- > No solvent use
- Lower production cost
- > Better performing batteries
- Challenges in production speed needs

BATTERY MATERIAL CHARACTERIZATION





Raw Material

- BET Surface area
- Pore size
- Pore volume
- Particle size
- Particle shape
- Skeletal density Tapped density
- Powder Rheology Defect ratio
- Crystallinity



Slurries

Flow behavior

- Slurry viscosity
- Zeta potential
- Density
- Percent solids
- Solvent concentration

Particle size

Particle shape

555 555

Drying

- Adhesion
- Mechanical properties
- stability Mechanical properties

Coating

Flow behavior

Slurry viscosity

Sedimentation

Calendering

- Hardness & elastic modulus
- Adhesion
- Mechanical properties



Stacking

- · Separator pore size
- Separator pore
- volume Mechanical properties



Battery Cell

- In operando characterization
- In situ characterization
- viscosity · Electrolyte density
- · Electrolyte flash point

Electrolytes

· Separator pore

Separator pore

size

volume

Electrolyte







IMPORTANT PHYSICAL PARAMETERS OF BATTERY RAW MATERIALS



BET SURFACE AREA



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- > Surface area of electrode material directly influences charge transfer between active materials and electrolyte
- > Surface area is optimized for electrical performance and to minimize adverse chemical reactions
- > Specific surface area of electrode materials is determined by the cryogenic adsorption of a gas (such as N2)
- Method uses gas sorption data to quantify the number of molecules of adsorbate in a monolayer on the surface of the material



PORE SIZE AND PORE VOLUME



 $N_{\rm 2}$ isotherm on 3 types of hierarchical carbon samples (mesopores)



- dVidi - Cumulative Proce Volu C1 0.5 - dV(d) - Cumulative C2 **NOVA Series** Kaomi Software - dV(d) - Cum C3 0.5

Merged CO₂ (273 K) and N₂ (77 K) DFT pore size distributions (blue) and cumulative pore volumes (red) for three carbons

 $\ensuremath{\text{CO}}_2$ isotherm on 3 types of hierarchical carbon samples (micropores)



PARTICLE PARAMETERS*

Si	ze	Size distribution		Shape	
Smal	II size	Uniform	Broad	Single cryst	tal structure
Shorter Li-ion transport distance	Larger SSA	Better cycle performance	Higher energy density	Higher electrode packed density	Difficult to manufacture
Better rate performance	More side reactions	Higher production difficulty	Poor cell homogeneity	Better cycle performance	Higher cost

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PARTICLE SIZE





Intensity-weighted particle size distribution of carbon black



Particle size distribution of three different cathode materials





PARTICLE SHAPE

Measurement results



Particle shape and size distribution of NMC cathode material measured with a dynamic image analyzer using a liquid flow dispersion unit

Particle shape affects parameters such as:

- > Homogeneity of the slurry
- > Uniformity of the coating on current collectors

For electrode material, near-spherical shape is preferred



Litesizer DIA

SKELETAL DENSITY



> Greater skeletal density of electrode material \rightarrow higher energy density of batteries



Principle of gas pycnometry:

Instead of weighing the displaced liquid, gas volumes are assessed via pressure measurements before and after expansion into an empty volume



Increase of skeletal density with increasing thermal treatment of nickel oxide electrode material



TAPPED DENSITY

- > Tapped density, also known as bulk or packing density, is an important parameter in battery material characterization, and incorporates:
 - > Volume of the solid phase
 - > Volume of pores within particles
 - > Volume of the spaces between particles
- For solid battery material, a greater tapped density produces more efficient energy management

Example: LiNiCoMnO₂ cathode material, mixed particle sizes results in greater tapped density and better electrochemical performance than with single particle size systems







POWDER RHEOLOGY



Compressibility of Si and Si/C samples as a function of normal stress

- > Si/C is denser than Si (also confirmed by gas pycnometry)
- Si almost completely lacks compressibility (densification)
- > Si powder readily densifies under its weight; it is very flowable => it acts as a flow-aid



MCR 302e / Powder shear cell



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Powder flow parameters of Si/C samples as a function of moisture

- Small effect of moisture levels between 0 % 50 %
- For moisture levels >50 % reduction in flowability, due to formation of menisci between particles
- > Important information for storage and transportation conditions

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DEFECT RATIO



Raman spectra of graphene powder (black) and a graphene film (red).



- "Graphitic" or "G" band at 1580 cm⁻¹: sharp signal in ordered graphite compared to amorphous carbon
- "Defect" or "D" band around 1350 cm⁻¹: signal associated with the breathing vibrations of the carbon atoms, becomes visible around defects
- The overtone of the D band, commonly referred to as the "2D" band, around 2680 cm⁻¹: for symmetry reasons, the 2D band in pure single-layer graphene is a perfect single-component peak
- > "Defect ratio" is the ratio of the D and G band intensities (I_D / I_G) that changes with the introduction of defects. Another ratio is the one between the G and the 2D band intensities (I_{2D} / I_G) which changes with the number of layers

DEGREE OF CRYSTALLINITY





Diffractogram of a graphite anode material

>	Powders with high degree of crystallinity facilitate electron
	mobility in the battery cell

- XRD gives information on the crystal structure, crystallite size, and phase purity of battery raw materials
- XRD is capable of detecting minor impurities when comparing the diffractogram of a known material with that of a contaminated one
- > Specialized gas-tight sample holders also make it possible to measure air-sensitive materials without sample degradation

Crystallite size (nm)	Degree of graphitization (%)
41.2	92.7





Sample holders for air-sensitive battery materials

XRDynamic 500



IMPORTANT PHYSICAL PARAMETERS OF BATTERY SLURRIES



FLOW BEHAVIOR



Speed	Torque	Temperature	Viscosity
[rpm]	[%]	[°C]	[mPa·s]
60	82.4	40	5945

Dynamic viscosity of slurry with single point measurement

The flow index is a description of the Newtonian or non-Newtonian behavior of the sample. A flow index that is smaller than 1 indicates that this slurry is shear-thinning – as is the case here for this slurry.



Calculated Herschel-Bulkley parameters to determine the flow behavior of slurry



ViscoQC 300

Mathematical Model Herschel-Bulkley

SLURRY VISCOSITY



Viscosity curves measured after stepwise addition of components, representing the processing steps for mixing procedures A and B



Comparison of mixing procedures A and B in terms of total solid fraction and viscosity, at constant shear rate



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ZETA POTENTIAL (ζ)



- > The zeta potential describes the stability of colloidal dispersions.
- > Zeta potential is an important parameter for the characterization of particles in infusions and emulsions.
- > The higher the magnitude of the zeta potential (highly + OR highly), the more stable the colloid.
- > Zeta potential is measured by electrophoretic light scattering (ELS)
- > During an ELS measurement an electric field is applied, which induces a collective motion of particles: particles with a higher zeta potential will move faster than less-charged particles.





Litesizer 500 Dynamic light scattering (DLS) Electrophoretic light scattering (ELS)

DENSITY AND PERCENT SOLIDS



> This is relevant for optimizing the surface area of the solid electrode



% solids formulated	Measured density (g/cm ³)	% solids calculated
41.45	1.3459	41.52
33.9	1.2682	34.27
26.30	1.1977	26.86
17.75	1.1226	17.96
15.22	1.1001	15.05
10.57	1.0688	10.81
6.41	1.0388	6.50
2.52	1.0121	2.45

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Formulated and calculated % solids in slurries with measured densities

Ultrapyc Series



SOLVENT CONCENTRATION

- > N-Methyl-2-pyrrolidone (NMP) is a strongly polar aprotic solvent commonly used in battery slurry preparation to dissolve binders like PVDF
- > NMP is also used as a solvent for lithium salts in electrolytes
- Because of its relative high price as well as its negative environmental impact, NMP is often recovered after use in the manufacturing process of lithium batteries
- Refractive index correlates with the concentration of NMP, so it is a quick and inexpensive way to measure it from 0.0% to 99.9%



Refractive Index of different NMP concentrations measured at 20 °C and 589 nm



Abbemat 550

SEDIMENTATION STABILITY



Frequency sweeps of sample battery slurry A and battery slurry B.



Schematic profile of the applied deformation during a frequency sweep test

- > A slurry is considered stable when the storage modulus G' is higher than the loss modulus G"
- > High frequencies enable evaluation of the samples' short-term behavior, while low frequencies simulate the behavior of the sample over a long time
- > G">G' would be considered the cross-over point where the slurry sediments
- In this example, slurry B exhibits better stability (cross-over at ~ 1 rad/s) compared to slurry A (cross-over at ~ 10 rad/s)



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View On-Demand Recordings in the Covalent Academy





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Thank you.